

Tellurium transport in the primary circuit of a nuclear power plant (NKS-R TETRA)

Anna-Elina Pasi NKS Seminar 24.-25.5.2022

WHY TELLURIUM?



- Releases from Fukushima and Chernobyl comparable to those of iodine and cesium
- Tellurium is one of the most significant radionuclides released during a nuclear accident
 - Volatile
 - Relatively long-lived isotopes
 - Somewhat chemically toxic
 - Decays to iodine
- So far, the release and behavior of tellurium in the core are well known but the transport to and behvaior in the containment are unclear

Experimental SA chemistry

- Experimentalists are a rare species nowadays
- Large facilites used in e.g. PHEBUS, VERCORS expensive
- Small-scale and detailed experiments still needed
- Managements systems like pool scrubbing and re-volatilization studies popular
- Important in developing codes and incorporating chemistry





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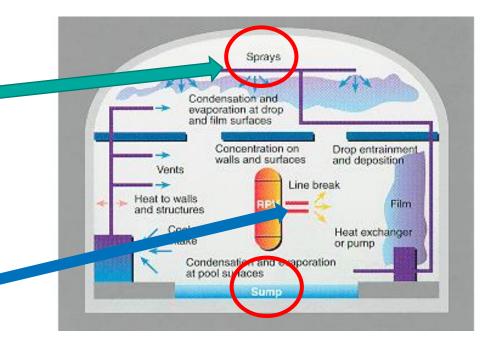
Focus of TETRA project



Removal of tellurium species by the containment spray system

Transport of tellurium species in the

Reactor Coolant System (RCS)



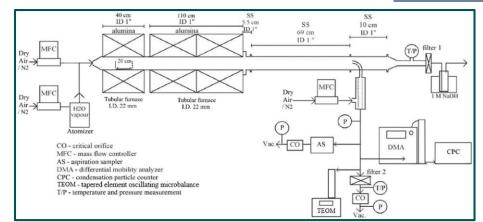
http://www.nucleartourist.com/systems/cs.htm

Experimental: Transport of Tellurium



How does atmosphere, humidity and other fission products affect tellurium transport?

- Six experiments conducted with changing parameters
- Air and nitrogen used as carrier gases
- Temperature chosen to represent RCS conditions
- Aerosol size and amount, particulate/gaseous species and precursor were analyzed



Condition	Experiment	Gas	Temperature ^a [K]	Humidity [ppmV]	Caesium iodide [M] ^b
Oxidizing	1	Air	1500	Dry	
	2	Air	1500	$2.1 \cdot 10^4$	
	3	Air	1500	$2.1 \cdot 10^4$	0.15
Inert	4	Nitrogen	1500	Dry	
	5	Nitrogen	1500	$2.1 \cdot 10^4$	
	6	Nitrogen	1500	$2.1 \cdot 10^4$	0.15

Temperature in the reaction furnace.

^b Concentration of the solution used in the atomizer.

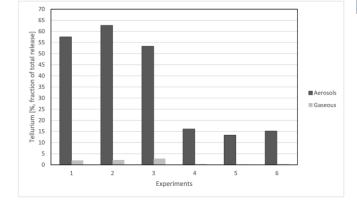
Results: Transport of Tellurium

- Majority of tellurium was transported in particualte form (aerosols), only small percentage as "gaseous"
- Humidity increased the amount Te in air → Supported by the literature
- Slight decrease in transport when Csl added in air
- Significantly lower release in inert (N₂)

No unexpected species observed

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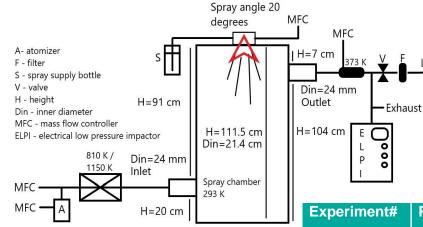
Condition	Experiment	Te- aerosols (filter)	Te- gaseous (liquid traps)	Identified species	Agglomerated chains
Oxidizing	1	58%	2.0%	TeO ₂	Thin dendritic
	2	63%	0.95%	TeO ₂	structure
	3	50%	0.71%	TeO ₂	(<100)
				TeO ₃ , CsI	
Inert	4	16%	0.99%	TeO ₂ , Te	Chains of
	5	13%	0.57%	TeO ₂ , Te	spherical
	6	15%	0.94%	TeO ₂ , Te,	particles (<50)
				CsI, Cs, I	-



Experimental: Containment Spray System

Liquid





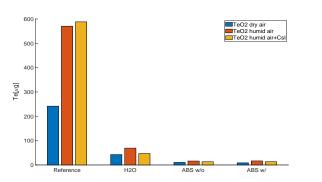
Spray solutions used:

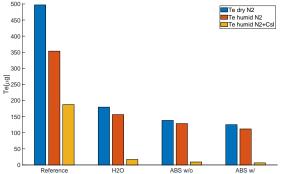
- Water
- Alkaline borate solution (ABS) without sodium thiosulfate
- Alkaline borate solution (ABS) with sodium thiosulfate

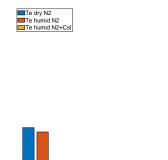
- Removal efficiency towards tellurium species formed under various conditions
- Effect of chemical composition of the spray

Experiment#	Precursor	Temperatur e, [K]	Atmosphere	Added humidity	Csl
1	TeO ₂	1150	Air	No	
2	TeO ₂	1150	Air	Yes	
3	TeO ₂	1150	Air	Yes	Yes
4	Те	810	Air	No	
5	Те	810	Air	Yes	
6	Те	810	Air	Yes	Yes
7	Те	810	Nitrogen	No	
8	Те	810	Nitrogen	Yes	
9	Те	810	Nitrogen	Yes	Yes

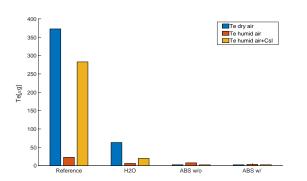
Results: Removal Efficiency











	Removal Efficiency [%]			
Experiment	MilliQ water	ABS without thiosulfate	ABS with thiosulfate	
1	83	96	97	
2	89	97	97	
3	91	97	97	
4	83	99	99	
5	74	73	73	
6	92	99	99	
7	63	72	75	
8	70	65	60	
9	89	93	94	

What can we learn form these results?



- Tellurium transport in the RCS is affected at least by atmosphere and humidity
 - Humidity increases the Te transport in air due to formation of more volatile species
 - Transport lower in N2 atmosphere
 - No new species formed in any of the experiments, transient/intermediate species possibly not detectable
 - Espegren, F., Kärkelä, T., Pasi, A. E., Tapper, U., Kučera, J., Lerum, H. V., ... & Ekberg, C. (2021). Tellurium transport in the RCS under conditions relevant for severe nuclear accidents. Progress in Nuclear Energy, 139, 103815.
- Containment spray system is efficient in removing tellurium species
 - Addition of CsI increased the removal due to agglomeration
 - Removal in nitrogen atmosphere was slightly lower \rightarrow Shift in particle size distribution
 - Chemical effect considered negligible
 - Kärkelä, T., Pasi, A. E., Espegren, F., Sevón, T., Tapper, U., & Ekberg, C. (2021). Tellurium retention by containment spray system. Annals of Nuclear Energy, 164, 108622.

Current and future work on tellurium

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- Tellurium behavior in the containment sump
- Organic tellurium chemistry \rightarrow Formation of volatile species
- Management of these possible re-volatilized species e.g. scrubber efficiency
- Gas phase tellurium chemistry, effect of organics \rightarrow NKS ORTEF project currently ongoing

Thank you for all partners, NKS, Nuclear Research Institute Rez! Thank you for for your attention!



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