

# Soaking foodstuff with lye as a counter measurement to reduce radioactivity.

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Sted, dd.mm.yyyy

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Statens strålevern  
Norwegian Radiation Protection Authority

# Introduction

- Different possibilities
- -block uptake with stable analogue isotope or stable isotopes (ex Iodine)
- -fertilizing land
- -let contaminated animals feed the last time on not contaminated food
- -use contaminated food as food for eel, crayfish etc with short biological half life and let them eat uncontaminated food the last time before "harvest"
- -liming lake and wetland
- -soaking with brine
- -salt stones with copperferrocyanate

# Soaking with lye

- **Old scandinavian tradition**
- -special occupation as "fiskblöteskor"
- -Mentioned by Carl Michael Bellman (1740-1795)
- *Se Dansmästaren Mollberg, Bröder,  
I vår Kröger-stuga,  
Hur mot väggen han sig stöder,  
Med en röd Fiol.  
Konstigt har han lärt sig buga,  
Och med foten skrapa.  
Aldrig såg man större Apa  
I en Capriol.  
Hjelp Himmel nå!  
Nu sprang han öfver Disken,  
Bums i en Så,  
Där Krögarmor har Fisken.  
Himmel ach! Se på hur luten  
Dryper af Surtouten.  
Nåsan tätt med blommor guten  
Skins som en Sol.*
- -Lutefisk was allowed to eat during the fast

# Material

- Ling from (*Molva molva*) GLADA FISKEN, Malmö
- Ling material from Skärhamns frys AB (Lutefisk factory), 2 kg of lutefisk, 25 l of lye solution, 25 l of rinse solution
- Pike (*Esox lucius*), from Ryssby lake, 20 km East Ljungby, 1.2-1.6 kg
- Reindeer fillet (*Rangifer tarandus*), Östersund market hall, 2 fillets 0.35 kg each.
- Reindeer steak and liver, Viltspecialisten Malmö. 1 kg steak, 0.7 kg liver

# Methods

- Soaking process
- -Using recipe from "Princessornas kokbok, 1938"
- -Swelling the fish for 1 week-changing water 4 times
- -0.1 kg of NaOH and 0.5 litre of  $\text{Ca}(\text{OH})_2$  + demieralized water. Sample/ water ration 1:5.
- -Potassium bicarbonate produced by boiling birch ash
- -One week soaking.
- -Rincing with water during one week, changing water every day
- -Temperature 10-15 °C
- -No swelling proces for pike and reindeer (fresh material)

# Measurements

- -Gamma spectrometry for  $^{137}\text{Cs}$  and  $^{40}\text{K}$  after drying at  $80\text{ }^{\circ}\text{C}$
- -For pike and ling skin and bone were removed
- -Solutions were evaporated to 180 ml or co-precipitated with copper ferrocyanate using  $^{134}\text{Cs}$  as yield determinant
- $^{210}\text{Po}$  was measured by alpha spectrometry after wet ashing with nitric and perchloric acid and deposited on Ni-discs



# Results

<u>Ling</u>	<u><math>^{137}\text{Cs}</math>(Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
Before treatment	2.42	275	238
After treatment	0.04	-5	2030
Reduction level	98%	100%	increase



# Ling from Skärhamns frys AB

<u>Ling</u>	<u>137Cs (Bq/kg)</u>	<u>40K (Bq/kg)</u>	<u>210Po (mBq/kg)</u>
Before treatment (fillet)	2.96	426	316
After treatment	1.8	7.1	2101
Reduction level	40%	98%	increase



# Pike

<u>Pike</u>	<u><math>^{137}\text{Cs}</math> (Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
Before treatment	96	524	739
After treatment	19	47	465
Reduction	80%	91%	41%




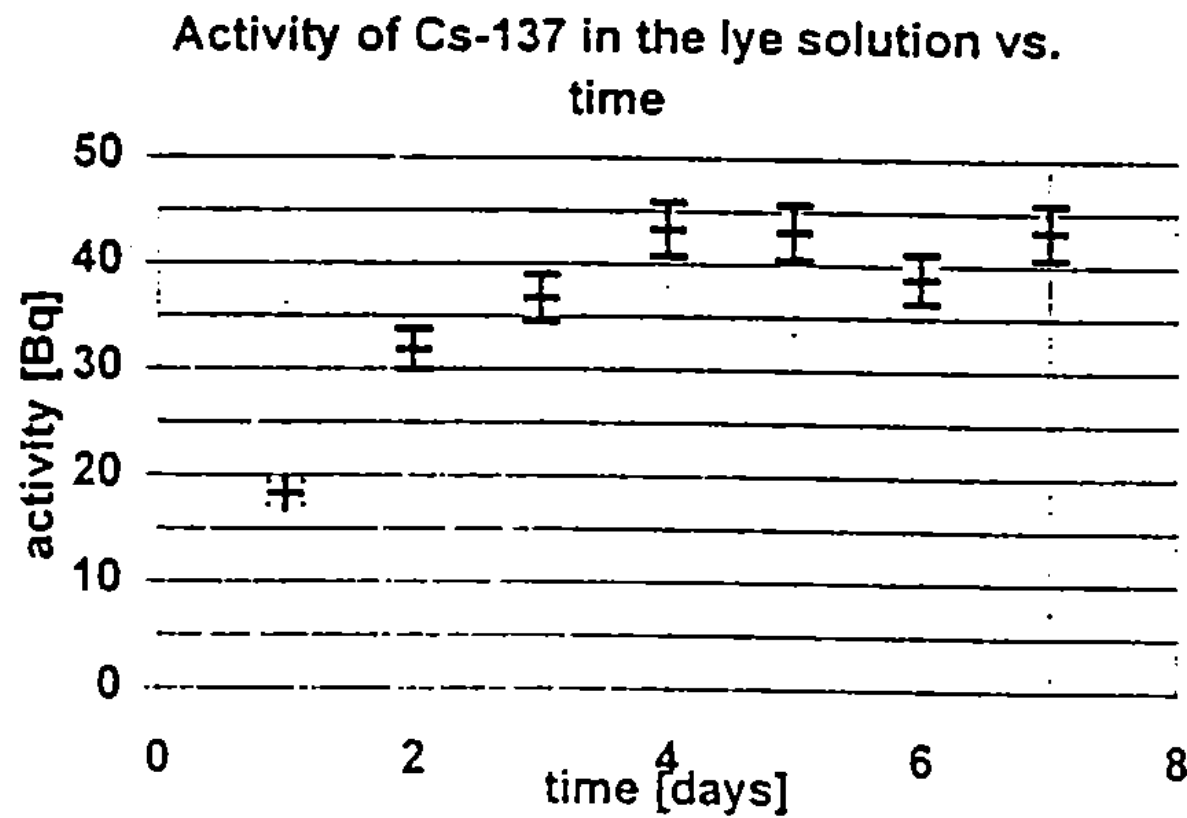
# Reindeer fillet

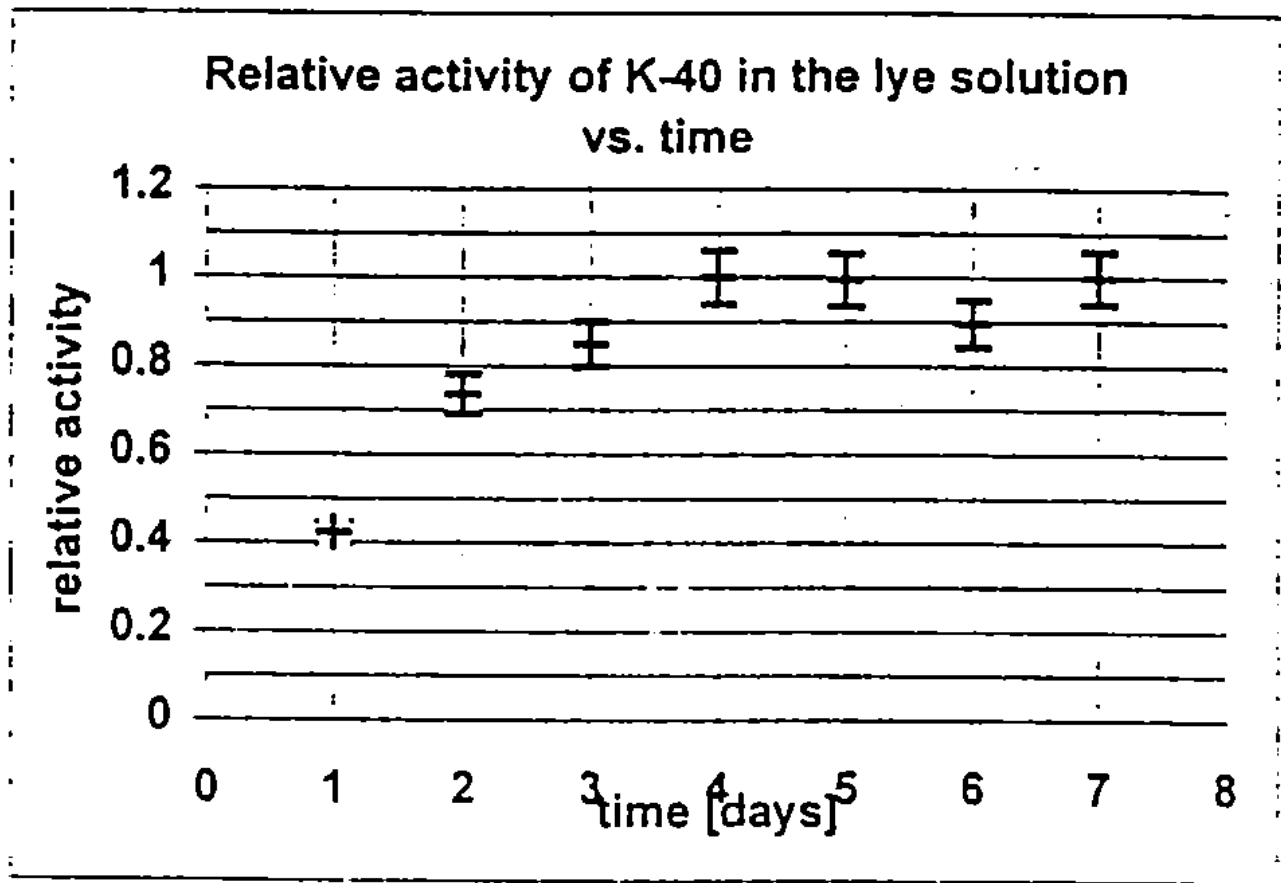
<u>Reindeer fillet</u>	<u><math>^{137}\text{Cs}</math> (Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
Before treatment	1825	386	8339
After treatment	349	1150	2495
Reduction	81%	increase	70%



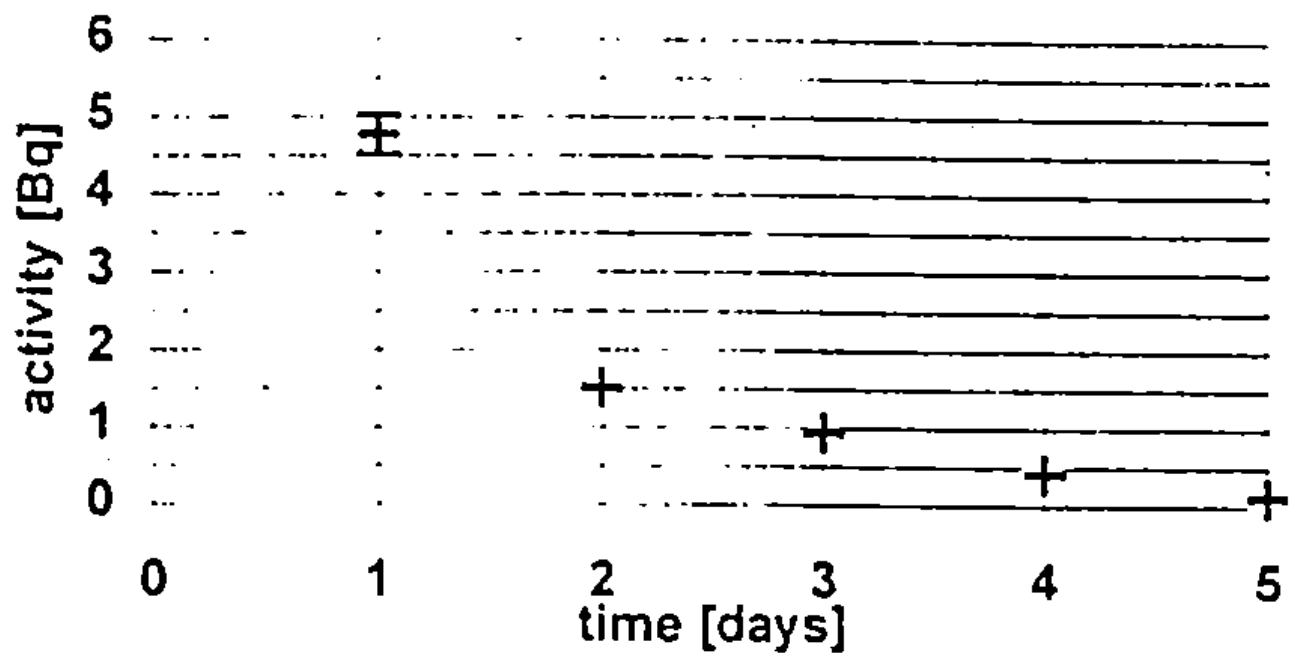
# Reindeer steak

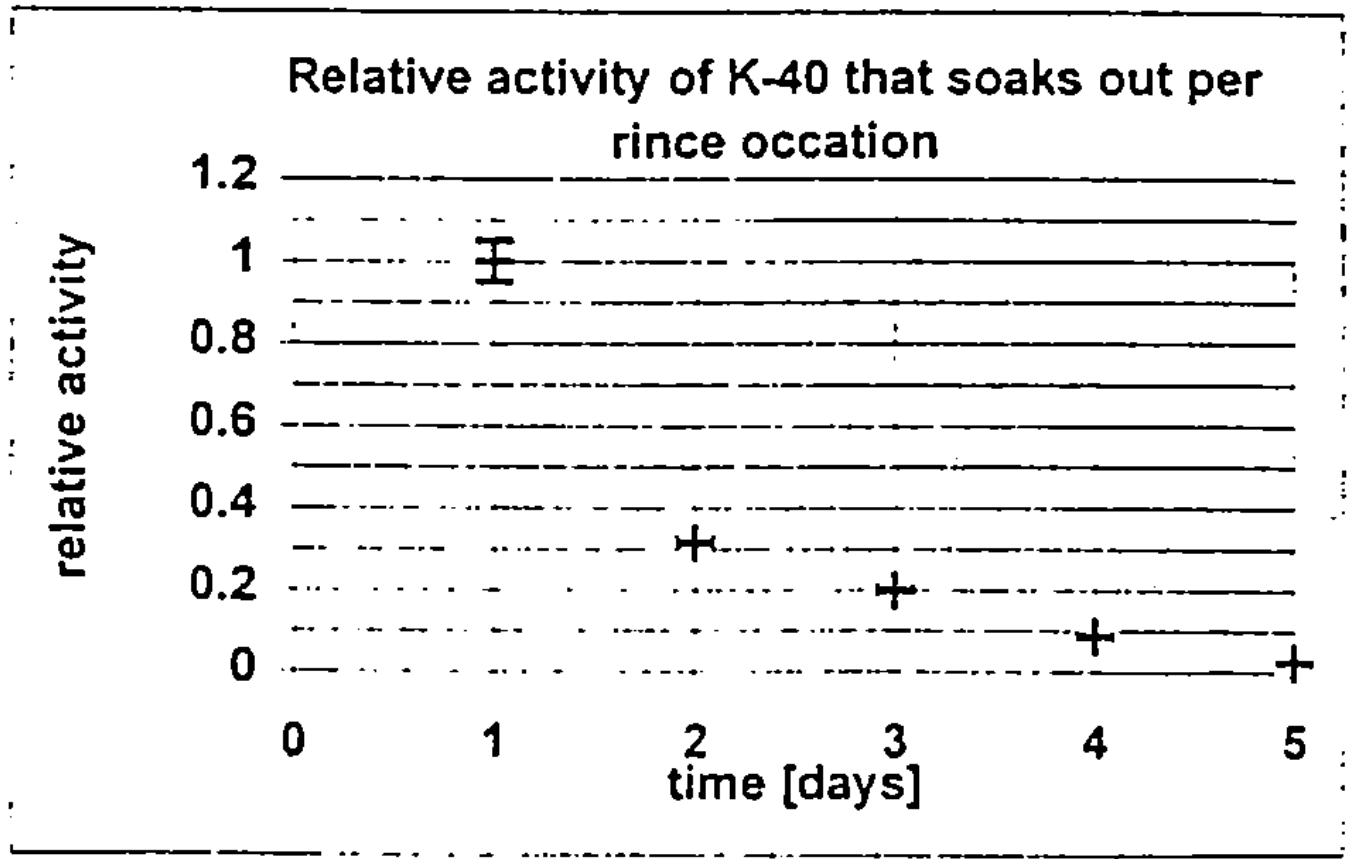
<u>Reindeer steak</u>	<u><math>^{137}\text{Cs}</math> (Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
1017	616		11140
19	NM		10286
98%			7.6%
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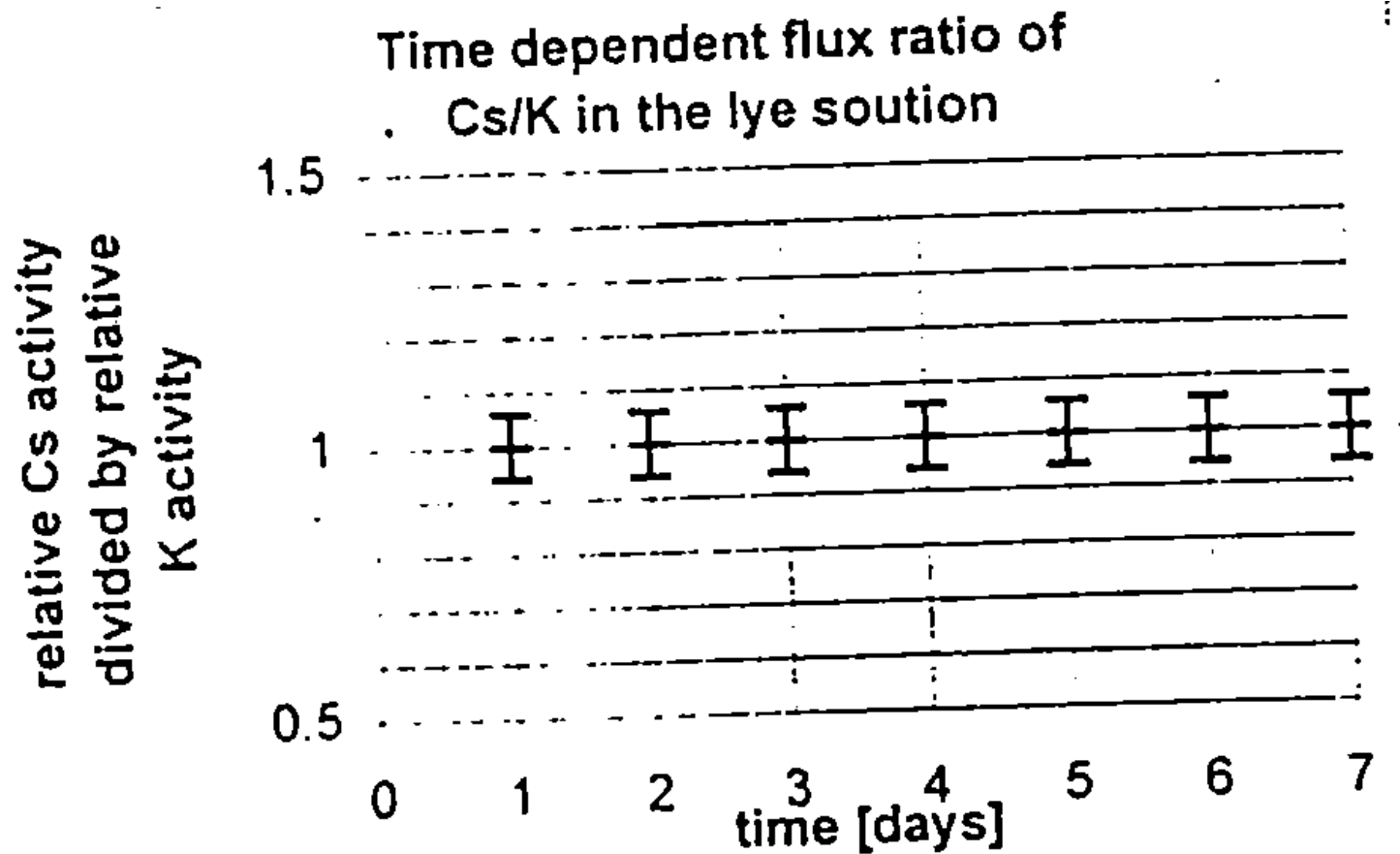




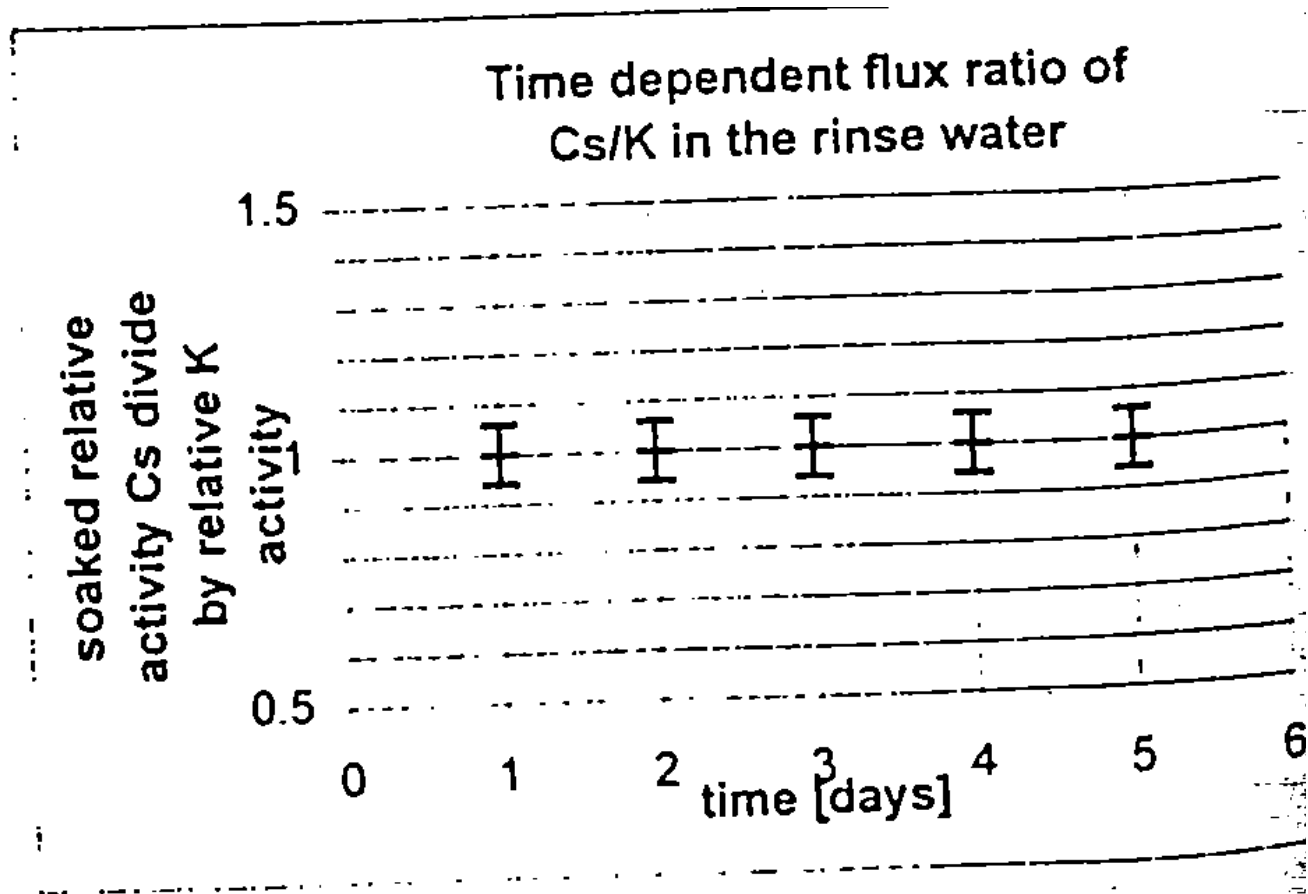
### Cs-137 activity that soaks out per rince occasion











# Reindeer-potash

<u>Reindeer steak</u>	<u><math>^{137}\text{Cs}</math> (Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
Before treatment	1017	616	11140
After treatment	196	477	8148
Reduction	81%	23%	27%



<u>Reindeer liver</u>	<u><math>^{137}\text{Cs}</math> (Bq/kg)</u>	<u><math>^{40}\text{K}</math> (Bq/kg)</u>	<u><math>^{210}\text{Po}</math> (mBq/kg)</u>
Before treatment	75	1763	8615
After treatment	13	388	10330
Reduction	82%	78%	increase



# Conclusions

- It is an effective method to reduce  $^{40}\text{K}$  and  $^{137}\text{Cs}$  in foodstuff (80-100%)
- There is an high intracellular concentration of potassium compared to extracellular.
- There is a breakdown of the cell membrane by NaOH and the alkali ions can move freely.
- The breakdown of proteins is a slower process.
- Effective for alkali metals. The cell membranes are broken down and the ions can move into the lye solution
- Cs behaves as K
- From counter measurement point of view use thin pieces of meat and one day soaking so cell membranes are broken down.
- Less effective for Po. Po bound to proteins.
- NaOH more effective than Potash
- It is from radiological point of view safe to eat soaked food stuff with lye. But the food is terrible



